BT / CSE / ECE / EEE 113

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I/IV B. TECH. DEGREE EXAMINATIONS, NOV / DEC - 2015 First Semester

BT / CSE / ECE / EEE

ENGINEERING CHEMISTRY-I

Time : **Three Hours** Maximum Marks : **60**

Answer Question No. 1 Compulsory. 12x1=12 M
Answer ONE question from each Unit. 4x12=48 M

- 1. Answer the following.
 - a) What is meant by Hard Water?
 - b) Scale & Sludge formation in boilers.
 - c) Units of hardness.
 - d) Secondary cells.
 - e) Solid state & Molten solvent batteries.
 - f) Nanostructured electrode materials.
 - g) Effect of pH on Corrosion.
 - h) Differential aeration Corrosion.
 - i) Galvanic series.
 - j) Functionality.
 - k) Drawbacks of natural rubber.
 - 1) Copolymerization.

UNIT - I

- 2. a) Explain the determination of hardness of water by EDTA method.
 - b) Explain softening of water by ion exchange process.

(OR)

- 3. a) Explain drinking water treatment in detail.
 - b) What are the WHO Limits of drinking water.

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UNIT - II

4. a)	Write a note on fuel cells.
b)	What are the scientific prospects of fuel cells?
	(OR)
5. a)	Describe Li ion batteries and their working.
b)	What are their applications?
	UNIT - III
6. a)	Explain the mechanism of dry Corrosion.
b)	Write an account on sacrificial anodic method.
	(OR)
7. a)	What are the Corrosion inhibitors? Explain the mechanism of inhibition.
b)	Write a note on
	(i) Galvanic Corrosion.
	(ii) Pitting Corrosion.
	UNIT - IV
8. a)	Explain the preparation and properties of the following
	(i) Bakelite.
	(ii) PVC.
b)	What are conducting polymers? What are their applications?

9. a) Distinguish between addition & condensation polymerization.

b) Write a note on Buna-S & Buna-N.

(OR)